

Note d'application

Improving the Speed and Quantitative Performance for the Separation of Furans and the Passivator Benzotriazole in Transformer Oil Analysis

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Abstract

Two fast, selective, and sensitive methods have been developed for the analysis of furans and the passivator benzotriazole in transformer oil. One method utilizes the ACQUITY UPLC H-Class System coupled to the ACQUITY UPLC PDA Detector for the rapid four-minute analysis of furans alone. The second method utilizes the ACQUITY UPLC H-Class System coupled to Xevo TQ MS if furans and additional transformer oil additives are required, the passivator benzotriazole was considered within this application.

Benefits

This application note illustrates increased sample throughput for the identification and quantification of furans and the passivator benzotriazole in transformer oil offering:

- Reduction of solvent usage due to reduced run times.
- Improved sensitivity, selectivity, and robustness, compared with existing methodologies.

Introduction

Transformer oil is usually a highly-refined mineral oil and due to its stability at high temperatures, it has excellent electrical insulating and heat transfer properties. Because of these properties transformer oil has many uses including: oil-filled transformers, high-voltage capacitors, fluorescent lamp ballasts, high-voltage switches, and in circuit breakers.

Furanic compounds can originate from the thermal depolymerization of cellulose solid insulation used within electrical equipment, which can then migrate into the transformer oil.

Electrical and mechanical stresses and the presence of water and oxygen increase the degree of cellulose degradation, eventually rendering the transformer ineffective. These changes can occur either due to normal aging or the presence of a fault.

Therefore an effective preventative maintenance program is critical to aid the planning of system servicing and, as a result, reduce costly instrument downtime. Periodic analysis of transformer oil for furans can be used to assess the degree of depolymerization of the cellulose, which does not require taking the unit out of service in order to take a sample.

Various additives can be incorporated into transformer oil to boost the quality of the oil, for either short or long term. All additives present need to be declared, therefore methodology to confirm the presence of declared additives, for example the passivator benzotriazole (BTA) – but also to detect undeclared additives, for example passivator Irgamet 30¹ – could be very valuable.

The presence of corrosive sulfur in transformer oil could potentially cause expensive power transformer failures. Passivators² (metal deactivators), such as BTA, act as long term corrosion inhibitors when added to transformer oil. They react with reactive metal surfaces both on metal surfaces and with dissolved metals (copper and silver) therefore reducing the rate of reaction with other compounds in the oil such as corrosive sulfur. But there are passivators, for example Irgamet 30,¹ that are not declared but are added to transformer oils to temporarily boost the oils quality to aid the passing of acceptance testing, therefore giving a false indication about the quality of the transformer oil.

Experimental

Sample preparation

Transformer oil analysis

- 1-mL transformer oil was extracted with 2 mL of acetonitrile for 1 min using a vortex mixer.
- 100 µL of the acetonitrile layer were transferred to an LC vial and diluted with 900 µL water.

LC conditions

| | |
|---------------|-------------------------------------------------------|
| LC system: | ACQUITY UPLC H-Class |
| Runtime: | 4.00 min |
| Column: | ACQUITY UPLC BEH C ₁₈ 1.7 µm, 2.1 x 150 mm |
| Column temp.: | 40 °C |
| Sample temp.: | 10 °C |

Mobile phase A: Water:(0.1% formic acid)

Mobile phase B: Acetonitrile:(0.1% formic acid)

Flow rate: 0.45 mL/min

Injection volume: 10.0 µL

Mobile phase gradient is detailed in Table 1.

| | Time (min) | Flow rate (mL/min) | %A | %B | Curve |
|---|------------|--------------------|----|-----|-------|
| 1 | Initial | 0.450 | 80 | 20 | – |
| 2 | 1.50 | 0.450 | 80 | 20 | 6 |
| 3 | 2.00 | 0.450 | 60 | 40 | 6 |
| 4 | 2.01 | 0.450 | 0 | 100 | 6 |
| 5 | 3.00 | 0.450 | 0 | 100 | 6 |
| 6 | 3.01 | 0.450 | 80 | 20 | 6 |
| 7 | 4.00 | 0.450 | 80 | 20 | 6 |

Table 1. ACQUITY UPLC H-Class mobile phase gradient.

UV conditions (for furans)

UV system: ACQUITY UPLC PDA Detector

Range: 190 to 350 nm

Resolution: 1.2 nm

Sampling rate: 20 pts/sec

Filter time constant: Normal

MS conditions (for furans and the passivator benzotriazole)

| | |
|-------------------|------------------------------------|
| MS system: | Xevo TQ MS |
| Ionization mode: | APCI positive |
| Corona current: | 20 μ A |
| Source temp.: | 150 °C |
| APCI probe temp.: | 400 °C |
| Desolvation gas: | 1000 L/hr |
| Cone gas: | 100 L/hr |
| Acquisition: | Multiple Reaction Monitoring (MRM) |

Optimum UV absorbances were established for each furan, detailed in Table 2, for the analysis of furans.

| Chemical substance | | CAS number | Retention time (min) | UV optimum absorbance (nm) |
|-------------------------------|------|------------|----------------------|----------------------------|
| Furans | | | | |
| 5-Hydroxymethyl-2-furaldehyde | 5H2F | 67-47-0 | 1.02 | 284 |
| Furfuryl alcohol | 2FOL | 98-00-0 | 1.28 | 216 |
| 2-Furaldehyde | 2FAL | 98-01-1 | 1.45 | 277 |
| 2-Furylmethylketone | 2ACF | 1192-62-7 | 1.77 | 274 |
| 5-Methyl-2-furaldehyde | 5M2F | 620-02-0 | 2.13 | 292 |

Table 2. Furans, associated CAS numbers, expected retention times, and the UV optimum absorbances.

The standard method for the analysis of furanic compounds in electrical insulating liquids is ASTM D5837-12,³ which uses HPLC with UV detection. Other example methodologies for the analysis of furanic compounds include: UPLC with UV detection,⁴ HPLC analysis using various detection methods including ultraviolet (UV),⁵ mass spectrometry (MS),⁶ refractive index,⁵ spectrophotometric⁷ analysis, and gas chromatography-mass

spectrometry (GC-MS).

Many previously used methods for furans consider only one furan (2-furaldehyde⁶) and lack reproducibility, selectivity, and sensitivity: they also require lengthy, costly, and time-consuming pre-treatments (SPE, concentration stages).

This application note describes two methods for the analysis of furans and the passivator benzotriazole in transformer oil. One method utilizes Waters ACQUITY UPLC H-Class System coupled to an ACQUITY UPLC PDA Detector for the rapid four-minute analysis of furans alone. The second method utilizes Waters ACQUITY UPLC H-Class System coupled to Xevo TQ MS if the passivator benzotriazole analysis is also required.

MS conditions were also optimized, detailed in Table 3, for the analysis of furans and the passivator benzotriazole. The established MRM method is illustrated in Figure 1.

| Chemical substance | | CAS Number | Retention time (min) | Cone voltage (V) | Transition | Collision energy |
|-------------------------------|------|------------|----------------------|------------------|---------------|------------------|
| Furans | | | | | | |
| 5-Hydroxymethyl-2-furaldehyde | 5H2F | 67-47-0 | 1.05 | 20 | 127.0 > 109.0 | 15 |
| | | | | | 127.0 > 81.0 | 20 |
| Furfuryl alcohol | 2FOL | 98-00-0 | 1.30 | 25 | 81.1 > 53.0 | 15 |
| 2-Furaldehyde | 2FAL | 98-01-1 | 1.46 | 25 | 97.1 > 69.0 | 15 |
| | | | | | 97.1 > 41.1 | 15 |
| 2-Furylmethylketone | 2ACF | 1192-62-7 | 1.79 | 20 | 111.1 > 43.0 | 20 |
| | | | | | 111.1 > 69.1 | 15 |
| 5-Methyl-2-furaldehyde | 5M2F | 620-02-0 | 2.15 | 25 | 111.1 > 55.0 | 20 |
| | | | | | 111.1 > 83.0 | 15 |
| Passivator | | | | | | |
| Benzotriazole | BTA | 95-14-7 | 1.66 | 40 | 120.1 > 65.0 | 20 |
| | | | | | 120.1 > 92.0 | 20 |

Table 3. Furans and the passivator benzotriazole, associated CAS numbers, expected retention times, cone voltages, MRM transitions, and associated collision energy values.

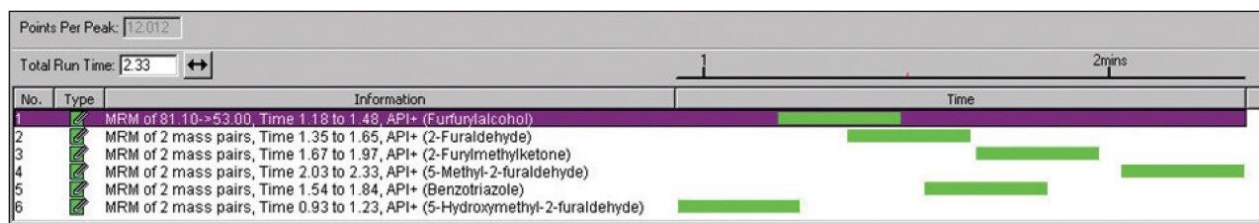


Figure 1. Furans MRM method (six compounds).

Instrument control, data acquisition, and result processing

Empower 3 Software was used for the UPLC-UV method.

MassLynx Software v.4.1 and TargetLynx Application Manager were used for the UPLC-MRM method.

Results and Discussion

Furans

The analysis of five furans was achieved using an ACQUITY UPLC H-Class System with an ACQUITY UPLC PDA Detector.

Optimum UPLC and UV conditions were developed, with the elution of all compounds within a four-minute run. When compared to the standard ASTM D5837-122 method with a minimum run time of 28 min, greater than seven times increase in sample throughput was achieved, with an excess of 88% less solvent usage.

Matrix matched mixed calibration standards, 0.01 to 5.00 µg/mL, were prepared and analyzed for all the compounds considered (equivalent range of 2 to 100 mg/kg in transformer oil using the developed method).

The calibration curve results generated by Empower Software for 5-hydroxymethyl-2-furaldehyde are shown in Figure 2, and the UV chromatograms, including associated UV spectra for each compound, are shown in Figure 3.

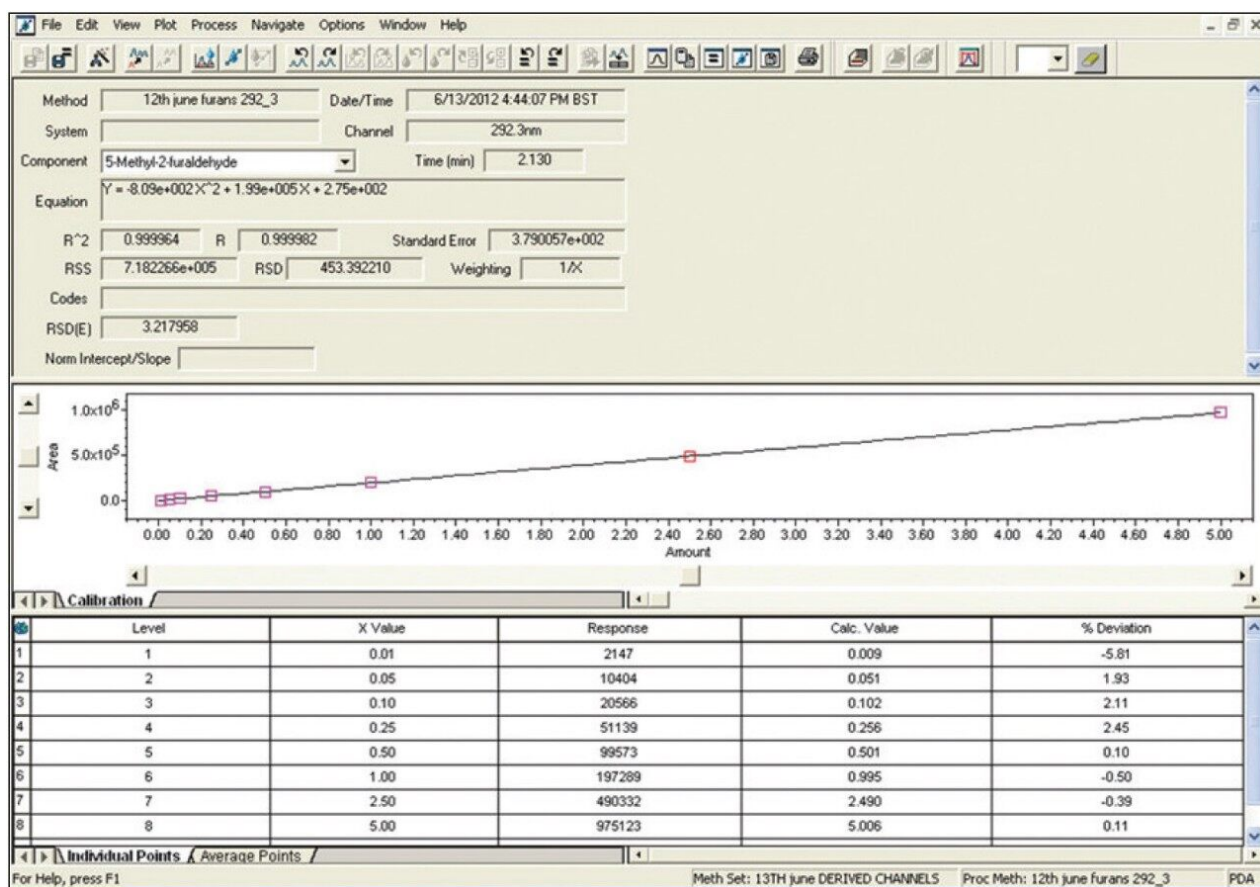


Figure 2. Empower Software calibration curve for 5-hydroxymethyl-2 furaldehyde.

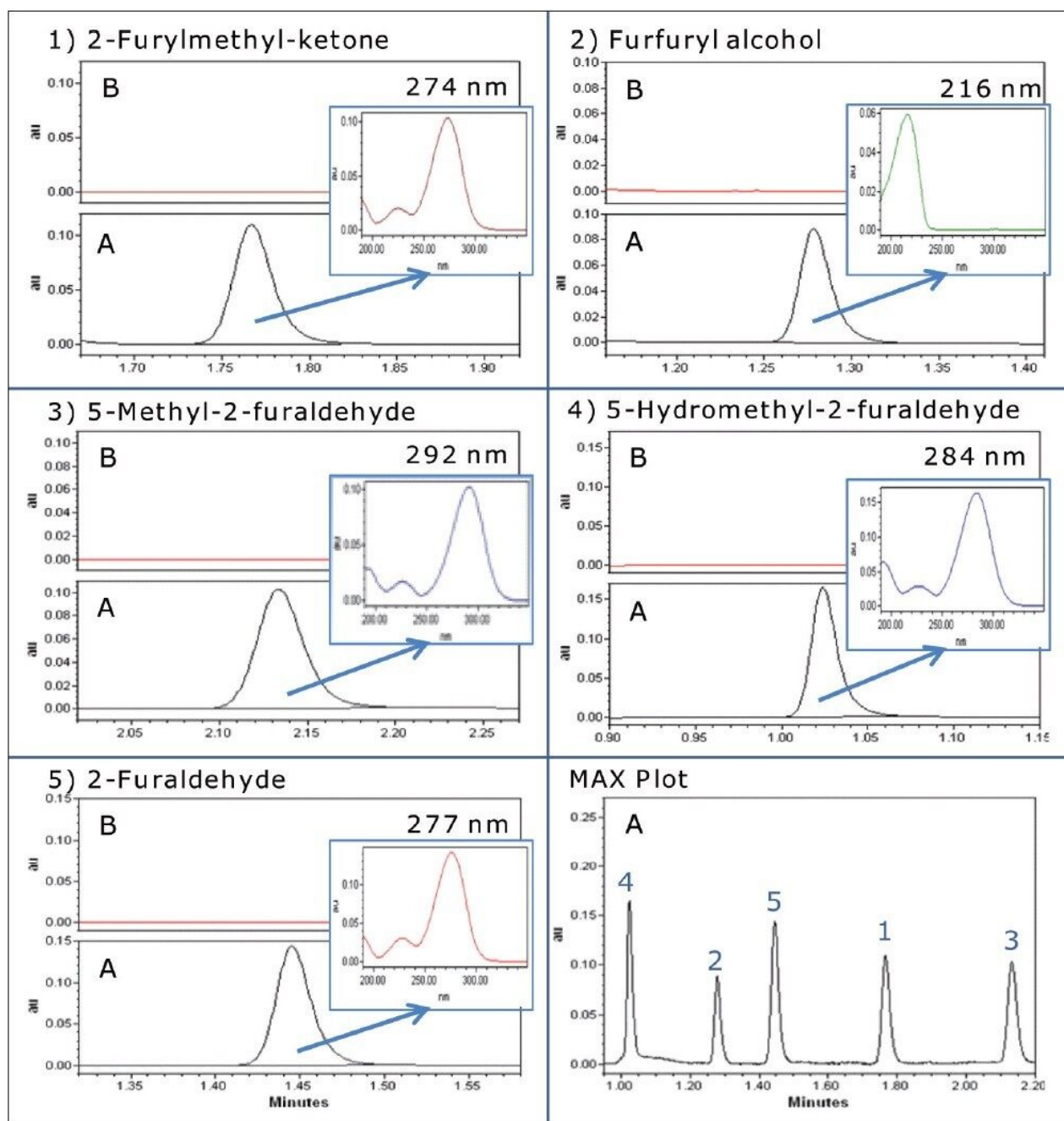


Figure 3. UV chromatograms and associated UV spectra for each furan in A) a mixed matrix matched $1 \mu\text{g/mL}$ calibration standard (equivalent to 20 mg/kg in transformer oil), and B) blank transformer oil.

Transformer oil analysis

Using the transformer oil analysis extraction and instrument parameters as detailed, the furan results obtained for the analysis of spiked transformer oil are detailed in Table 4. Recoveries were obtained by comparing extracted spiked transformer oil against matrix matched calibration standards.

| Compound | | UV results | | |
|-------------------------------|----------|----------------------------------------|-------|-------------------------|
| | | Replicate injection results (mg/kg) | | Average recovery (%) |
| | | 1 | 2 | |
| 5-Hydroxymethyl-2-furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.63 | 1.64 | 81.9 |
| | 10 mg/kg | 9.37 | 9.36 | 93.6 |
| Furfuryl alcohol | Blank | ND | – | – |
| | 2 mg/kg | 1.89 | 1.87 | 93.8 |
| | 10 mg/kg | 9.51 | 9.38 | 94.5 |
| 2-Furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.90 | 1.91 | 95.1 |
| | 10 mg/kg | 9.54 | 9.55 | 95.4 |
| 2-Furylmethylketone | Blank | ND | – | – |
| | 2 mg/kg | 1.94 | 1.95 | 97.3 |
| | 10 mg/kg | 9.59 | 9.58 | 95.8 |
| 5-Methyl-2-furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.86 | 1.86 | 93.2 |
| | 10 mg/kg | 9.96 | 10.01 | 99.8 |

Table 4. Transformer oil spiked with furans recovery data. Results obtained using PDA detection and quantified against matrix matched mixed calibration standards. ND = not detected.

Efficient recoveries were obtained, ranging between 82% to 100% for two replicates, no evidence of matrix interference was observed.

Furans and the passivator benzotriazole

The analysis of six compounds (five furans and the passivator benzotriazole) was achieved using Waters Xevo TQ MS in MRM mode with APCI ionization, coupled to an ACQUITY UPLC H-Class System.

Optimum UPLC and MRM conditions were developed, with the elution of all compounds within a four-minute run.

Transformer oil analysis

The MRM mass detection method detailed in Figure 1 was used after appropriate sample preparation to quantify for furans in transformer oil (containing furans and the passivator benzotriazole).

Using the transformer oil analysis extraction and instrument parameters as detailed, the furan results

obtained for the analysis of spiked transformer oil are detailed in Table 5. Recoveries were obtained by comparing extracted spiked transformer oil against matrix matched calibration standards.

| Compound | | MS results | | |
|-------------------------------|----------|-------------------------------------|-------|----------------------|
| | | Replicate injection results (mg/kg) | | Average recovery (%) |
| | | 1 | 2 | |
| Benzotriazole | Blank | ND | – | – |
| | 2 mg/kg | 1.97 | 1.95 | 98.1 |
| | 10 mg/kg | 10.55 | 10.27 | 104.1 |
| 5-Hydroxymethyl-2-furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.97 | 1.95 | 95.0 |
| | 10 mg/kg | 10.55 | 10.27 | 95.4 |
| Furfuryl alcohol | Blank | ND | – | – |
| | 2 mg/kg | 1.71 | 1.64 | 83.6 |
| | 10 mg/kg | 10.41 | 10.01 | 102.1 |
| 2-Furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.73 | 1.93 | 91.5 |
| | 10 mg/kg | 9.11 | 8.49 | 88.0 |
| 2-Furylmethylketone | Blank | ND | – | – |
| | 2 mg/kg | 1.74 | 1.75 | 87.4 |
| | 10 mg/kg | 9.90 | 10.10 | 100.0 |
| 5-Methyl-2-furaldehyde | Blank | ND | – | – |
| | 2 mg/kg | 1.99 | 2.09 | 102.0 |
| | 10 mg/kg | 10.56 | 10.58 | 105.7 |

Table 5. Transformer oil spiked with furans recovery data. Results obtained using mass spectrometric detection and quantified against matrix matched mixed calibration standards. ND = not detected.

Efficient recoveries were obtained, ranging between 84% to 105% for the two replicates. Additional benefits include improved selectivity and sensitivity for the analysis of furans and transformer oil additives (the passivator benzotriazole was considered within this application) using the Waters Xevo TQ MS in MRM mode.

Matrix matched mixed calibration standards, 0.01 to 5.00 µg/mL, were prepared and analyzed for all the compounds considered (equivalent range of 2 to 100 mg/kg in transformer oil using the developed method). The TargetLynx Quantify results for 5-hydroxymethyl-2-furaldehyde are shown in Figure 4, and the MRM

chromatograms for each compound are shown in Figure 5.

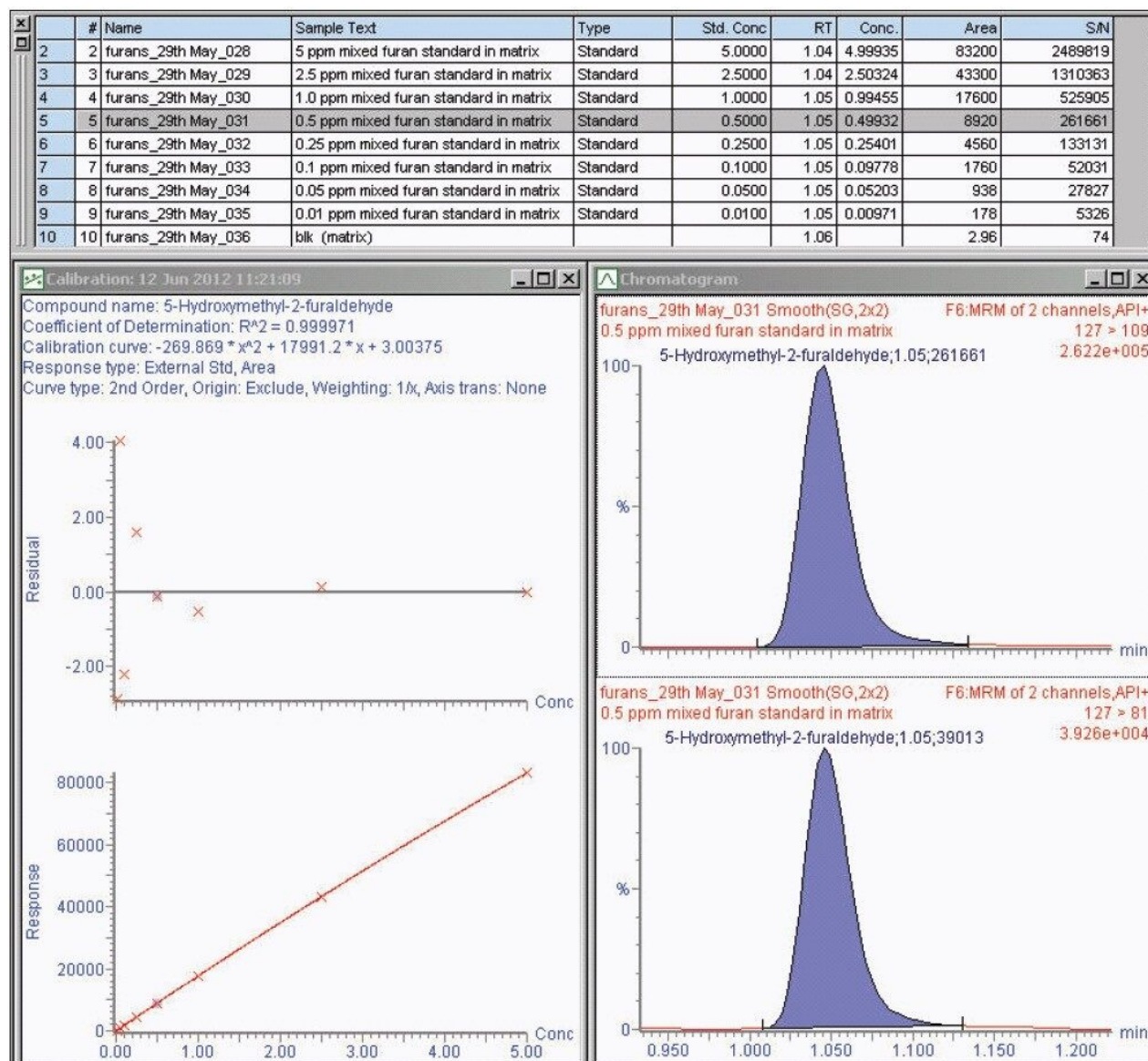


Figure 4. TargetLynx Quantify results browser shows the calibration quantitation results, calibration curve, and example MRM chromatogram for 5-hydroxymethyl-2-furaldehyde.

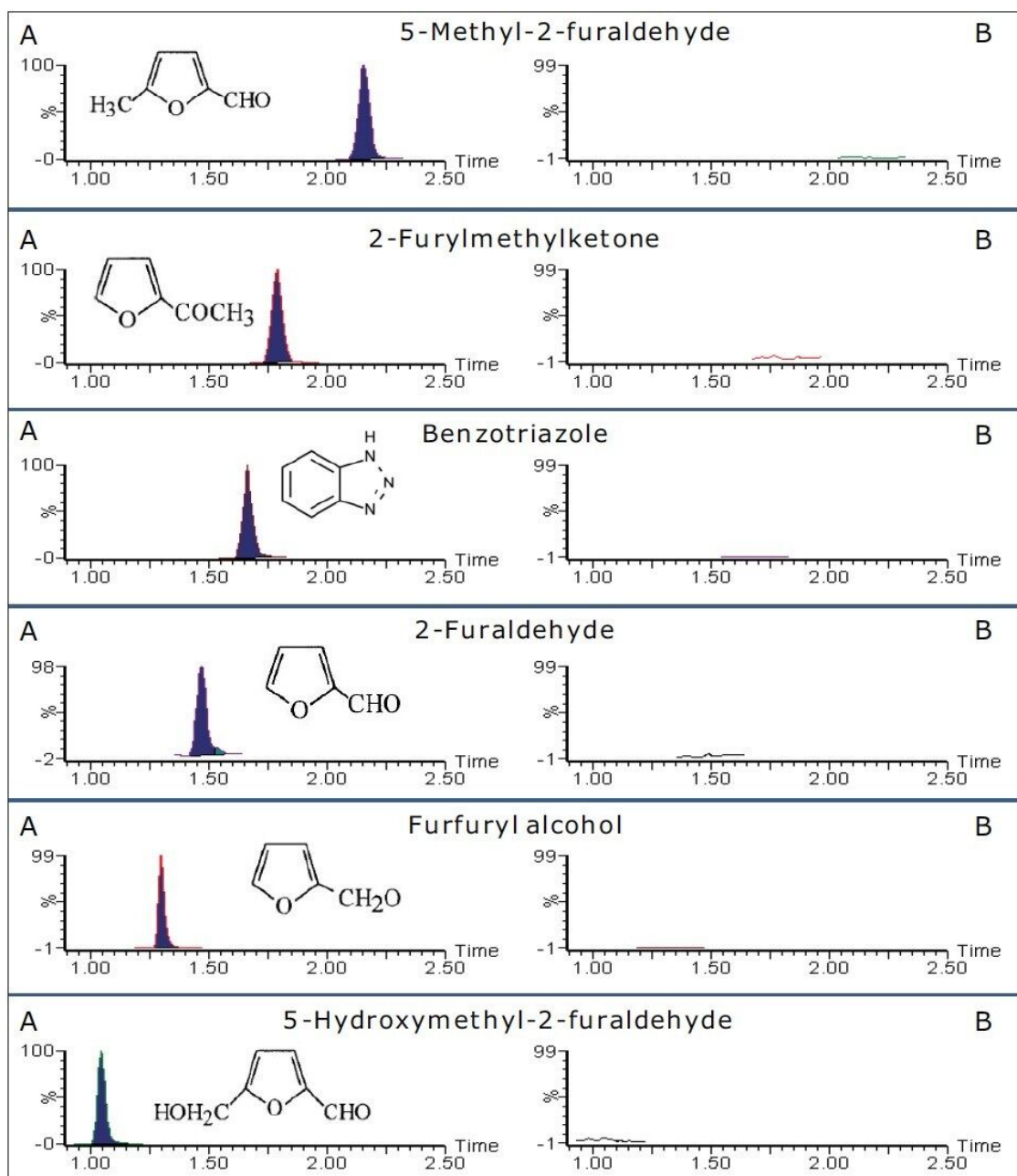


Figure 5. MRM chromatograms for furans and the passivator benzotriazole in A) a mixed matrix matched 1 $\mu\text{g/mL}$ calibration standard (equivalent to 20 mg/kg in transformer oil), and B) blank transformer oil.

Conclusion

Two fast, selective, and sensitive methods have been developed for the analysis of furans and the passivator benzotriazole in transformer oil.

One method utilizes the ACQUITY UPLC H-Class System coupled to the ACQUITY UPLC PDA Detector for the rapid four-minute analysis of furans alone. The second method utilizes the ACQUITY UPLC H-Class System coupled to Xevo TQ MS if furans and additional transformer oil additives are required, the passivator benzotriazole was considered within this application.

The described approach offers many benefits when compared to the standard method:

- Business benefits of using UPLC analysis including, greater than seven times increase in sample throughput and an excess of 88% less solvent usage.
- Enhanced sensitivity and selectivity resulting in improved confidence in identification and quantification offered using Xevo TQ MS.

APCI provides an efficient ionization mode for the analysis of the broad distribution of diverse polarities of the compounds being analyzed.

Quantitative furan analysis as detailed, using a small amount of insulating oil sampled from the transformer (without taking it out of operation), can be used as a representation of the degree of depolymerization and hence establish the condition of the transformer.

References

1. Laborelec GDF Suez. News: focus on electrical equipment. Journal twelfth year. No 124, April 2011. [cited 2012 July 12]. Available from http://www.laborelec.com/pages_files/news_april_11.pdf.
2. Lewand L. Passivators – What they are and how they work. Neta World, Spring 2006; [cited 2012 January 10]. Available from: <http://www.netaworld.org/files/neta-journals/NWsp06ChemPersp.pdf>
3. Standard test method for furanic compounds in electrical insulating liquids by high-performance liquid chromatography (HPLC). ASTM. D5837-12, Feb 2012.
4. Schwarz M, Rodriguez C, Gullillen D, Barroso C. Development and validation of UPLC for the determination of phenolic compounds and furanic derivatives in Brandy de Jerez. *J Sep Sci.* 32: 1782-1790.

5. Yuan J, Chen F. Simultaneous separation and determination of sugars, ascorbic acid and furanic compounds by HPLC – dual detection. *Food Chemistry*. 64: 423-427, 1999.
6. Koreh O, Ludanyi K, Vekey K, Javorszky E, Molnar E, Torkos K, Borossay J. Determination of furanic compounds in insulating oil by high performance liquid chromatography/mass spectrometry using atmospheric pressure chemical ionization. *Rapid Communications in Mass Spectrometry*. 12: 1515-1519, 1998.
7. Pahlavanpour B, Duffy G, Rapid spectrophotometry analysis of furfuraldehyde in transformer oil. Cigre 110-11, Symposium Berlin (1993).

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720004426, August 2012