



## Analysis of Glyphosate, Glufosinate, and AMPA in Tap and Surface Water Using Open-Architecture UPLC with 2D-LC Technology

---

Claude R. Mallet

Waters Corporation



---

## Abstract

The application targeted the analysis glyphosate, glufosinate, and AMPA in tap and surface water. The limit of quantification in this study was measured at 1.0 ppb. EPA regulations have an MRL in water set at 700 ppb; the 1.0 ppb quantification limit clearly meets these regulations.

### Benefits

- Automated derivatization protocol (15 min)
- Minimum sample pre-treatment (filtration only)
- Excellent retention with reversed-phase BEH C<sub>18</sub>

---

## Introduction

The popularity of glyphosate as a weed killer for crop protection is mainly due to its effectiveness against broadleaf plants. This herbicide acts as an enzyme inhibitor and is only active on growing plants. After absorption in soil, glyphosate is rapidly converted to its main metabolite (aminomethylphosphonic acid or AMPA). Due to its strong retention characteristic, it is not typically found in ground water, but can potentially contaminate surface waters through soil erosion and run-offs. Glyphosate's toxicity is classified at Level III by the EPA; as such, the herbicide is regulated to protect public health. Due to its ionic structure, poor volatility, and low molecule mass, the analysis of glyphosate in water at low ppb is very difficult.<sup>1</sup> Furthermore, the high polar nature, low volatility, and absence of chromophores are the prime reasons for the analysis and detection using a derivatized format<sup>2</sup> for herbicides. Several derivatization options have been evaluated and the ease-of-use approach of 9-fluorenylmethyl chloroformate (FMOC-Cl) for primary and secondary amines leads to a single multi-residue method for glyphosate and AMPA.<sup>3,4,5</sup>

The analysis of glyphosate in drinking water usually requires elaborate sample extraction and clean up protocol to minimize matrix effects. One major drawback is the high amount of manual labor required to produce a clean extract, leading to increased operator-induced error. Since glyphosate is highly soluble in water, a weak reversed-phase sorbent is usually used for enrichment purpose. Another drawback is the insolubility of glyphosate in other solvents (MeOH, IPA, ACN, acetone, etc). The analysis of glyphosate is further complicated by the low solubility of FMOC in water. From this point, the main challenge is to bring

the water-soluble analyte in contact with the organic-soluble (acetonitrile) derivatization agent (FMOC-Cl). This ultimately leads to a level of complexity regarding the ratio of water to organic solvent for optimum yield without causing a salting-out (glyphosate) or precipitation effect (FMOC). Also, with a high acetonitrile level present in the sample, potential breakthrough or peak distortion effect can be expected during separation.

LC-MS/MS and GC-MS/MS have been utilized for routine analysis since the introduction of hyphenated instrumentations in the 1970's. By improving the level of automation, the next generation of hyphenated solutions are even better equipped to bring a measurable cost reduction to the overall analytical process (time, resources, and consumables). Time de-coupled chromatography<sup>6</sup> offers automated sample handling and micro-extraction capabilities.

In this application, the analysis of glyphosate, glyfosinate, and AMPA in water was performed using three automated sequences for the derivatization and separation. The first part of the analysis performed the conversion of glyphosate and AMPA with the FMOC derivative. The second part of the analysis used an automated sequence for quenching the reaction. The final part of the analysis used an at-column dilution function for high-volume injection of the water:acetonitrile (66:33) sample. Up to 0.5 mL of derivatized sample was loaded onto a trap column. Several trapping sorbents were evaluated for trapping efficiencies. A weak reversed-phase sorbent gave the best performance. The trapped analytes were analyzed on a high resolution column using a back flush gradient. With this automated solution, glyphosate, glufosinate, and AMPA were detected at 1 ppb level (ug/L).

---

## Experimental

### Chromatography and MS/MS conditions

#### Loading Conditions

Column:	Oasis HLB 20 $\mu\text{m}$
Loading:	Water pH 7 no additives
Flow rate:	2 mL/min

## Loading Conditions

At-column dilution: 5% (0.1 mL/min pump A and 2 mL/min pump B)

## UPLC conditions

UPLC system: Open-Architecture UPLC 2D with at-column dilution

Runtime: 10 min

Column: ACQUITY UPLC BEH C<sub>18</sub>, 2.1 x 50 mm, 1.7 μm

Column temp.: 60 °C

Mobile phase A: Water + 0.5% formic acid

Mobile phase B: Acetonitrile + 0.5% formic acid

Elution: 5 min linear gradient from 5% (B) to 95% (B)

Flow rate: 0.5 mL/min (pump C)

Injection volume: 500 μL

## MS conditions

MS System: Xevo TQ MS

Ionization mode: ESI Positive

Capillary voltage: 3.0 kV

Cone voltage: 30.0 V

Source temp.:	150 °C
Desolvation temp.:	550 °C
Desolvation gas:	1100 L/hr
Cone gas:	50 L/hr

Two MRM transitions (quantification and confirmation) for glyphosate, glufosinate, and AMPA were selected and optimized. The MRM conditions are listed in Table 1 and the corresponding spectrums are shown in Figure 1. For this application, finding the optimum chromatographic condition for this multi-residue analysis poses a difficult challenge due to the chemical diversity. The chromatographic conditions were tested on several trapping chemistries (Oasis HLB, XBridge C<sub>18</sub>, and XBridge C<sub>8</sub>) and separation chemistries (BEH C<sub>18</sub> and HSS T3). The loading (low pH, high pH, and neutral pH) and eluting mobile phase (MeOH + 0.5% Formic acid; ACN + 0.5% Formic acid) were also optimized using an automated process. The derivatization protocol is listed in Table 2. Potassium borate and 9-fluorenylmethyl chloroformate (FMOC-Cl) were purchased from Sigma Aldrich. A 1-L pH 10 borate buffer (5%) was prepared and pH adjusted with ammonium hydroxide. The derivatization agent (FMOC-Cl) was prepared in 10 mL acetonitrile at 1.5 mg/mL concentration. Stock solutions of glyphosate, glufosinate, and AMPA were prepared in water at 1 mg/mL.

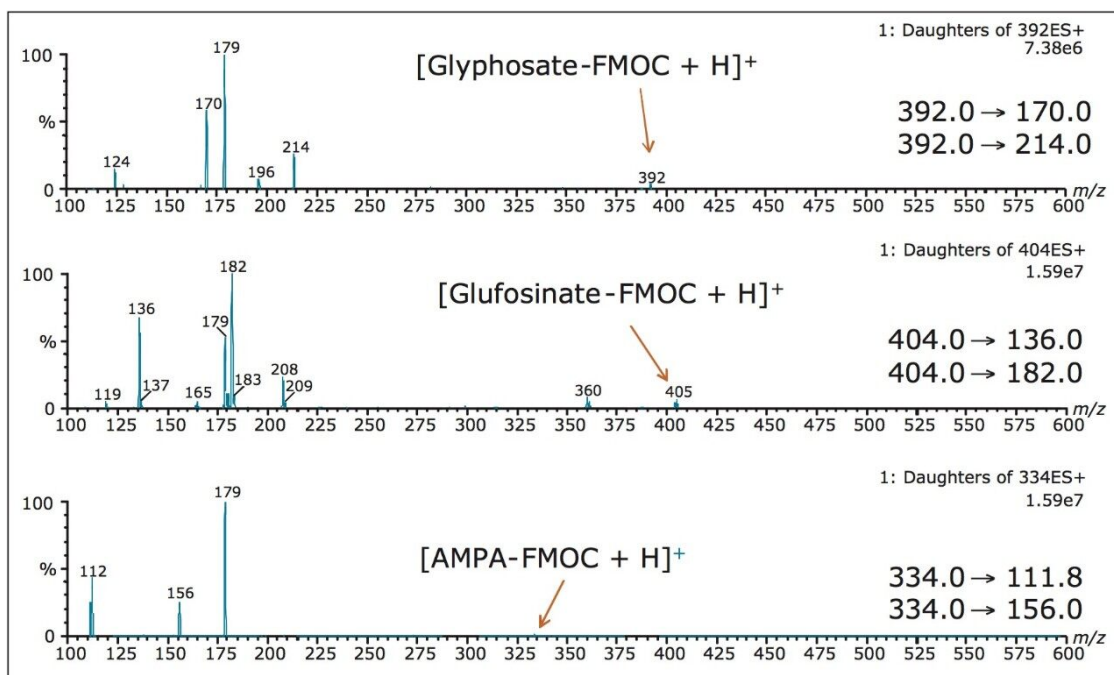


Figure 1. FMOc daughter spectrums for glyphosate, glufosinate, and AMPA.

Herbicides	Ion mode	Precursor ion	Cone	Product ion	CE
Glyphosate-FMOc	ESI+	392.0	20	170.0	15
				214.0	10
AMPA-FMOc	ESI+	404.0	15	136.0	15
				182.0	10
Glufosinate-FMOc	ESI+	334.0	20	111.8	20
				156.0	15

Table 1. MRM conditions for glyphosate, glufosinate, and AMPA.

<b>Step 1:</b>	10 mL water sample in 20 mL vial
<b>Step 1a</b>	Option – filter or clean up with anion/cation exchanger SPE
<b>Step 2:</b>	Add <b>250 µL</b> HCl pH 1 – binding effect reduction
<b>Step 2a:</b>	Option – Add <b>50 µL</b> Internal Standard
<b>Step 3:</b>	Add <b>5 mL</b> ACN – optimized derivatization conditions
<b>Step 4:</b>	Add <b>500 µL</b> Borate Buffer 5% pH 10 – neutral conditions
<b>Step 5:</b>	Add <b>500 µL</b> FMOCl (1.5 mg/mL)
<b>Step 6:</b>	30 min at 60 °C
<b>Step 7:</b>	Add <b>250 µL</b> HCl pH 1 – quench reaction – stable conditions
<b>Final sample composition:</b>	<b>2:1 Aqueous/ACN</b> (33% organic solvent)

Table 2. FMOCl derivatization protocol for glyphosate, glufosinate, and AMPA

The step-by-step derivatization protocol is shown in Figure 2. The procedure begins with 10 mL of filtered water sample. The first step starts with the addition of 250 µL of hydrochloric acid (1N) to release any binding effect. Next, the automated protocol aspirate 5 mL of acetonitrile and dispense the entire volume in the sample vial. At this point in the protocol, an internal standard can be added to the sample vial. The next sequence deals with the addition of the borate buffer and the derivative agent. This sequence used a high pH buffer (0.5 mL borate buffer pH 9) to neutralize the amine functionality, followed by the addition of the FMOCl derivative (0.5 mL). The reaction gave an optimum yield after 30 minutes at 60 °C temperature. The reaction is quenched and stabilized with the addition of 250 µL of hydrochloric acid (1N). At this point, the final sample composition is 2:1 aqueous:acetonitrile.

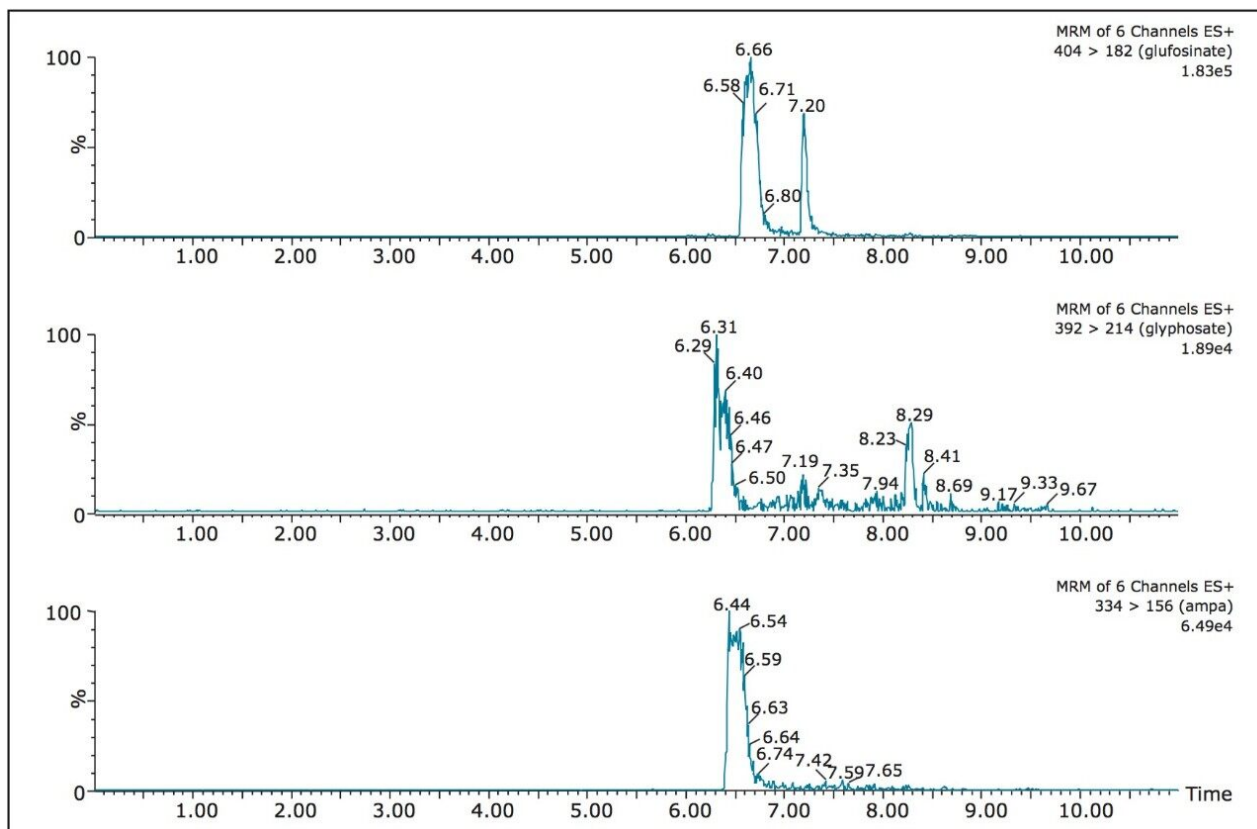


Figure 2. Chromatography profile of Fmoc-glyphosate, Fmoc-glyfosinate, and Fmoc-AMPA with at-column dilution inactive.

## Results and Discussion

### Automated method development

The starting point of any analytical protocol is the selection of chromatographic parameters to achieve well-resolved peaks for qualitative and/or quantitative analysis. Method development is typically performed with a trial-and-error approach, which ultimately leads to an optimized chromatographic method in a relatively short time. Another current practice is to select the most successful conditions in a systematic screening approach with the goal of quickly reaching optimized conditions. When utilizing multidimensional chromatography, the task of selecting optimized conditions can be quite difficult. However, with automation and a selection of key parameters, a large number of methods can be screened in a short time frame.<sup>6</sup>

For example, Figures 2 and 3 show the at-column dilution effect (ON and OFF) with optimized loading



conditions, elution conditions, trapping chemistries, and separation chemistries. As shown, with the at-column dilution inactive, the chromatography shows a wide peak shape for all three herbicides (Figure 2). The distorted peak shape for glyphosate, glufosinate, and AMPA are properly re-focused using a 5% at-column dilution, as seen in Figure 3. The derivatization process followed by an immediate analysis produced excellent reproducibility values in the 2% range (N=8). The FMOc derivative was found to be stable for 24 hrs. No further evaluation was performed to determine the stability limit, since the derivatization and analysis of 50 water samples can be process during an overnight run (15 hrs).

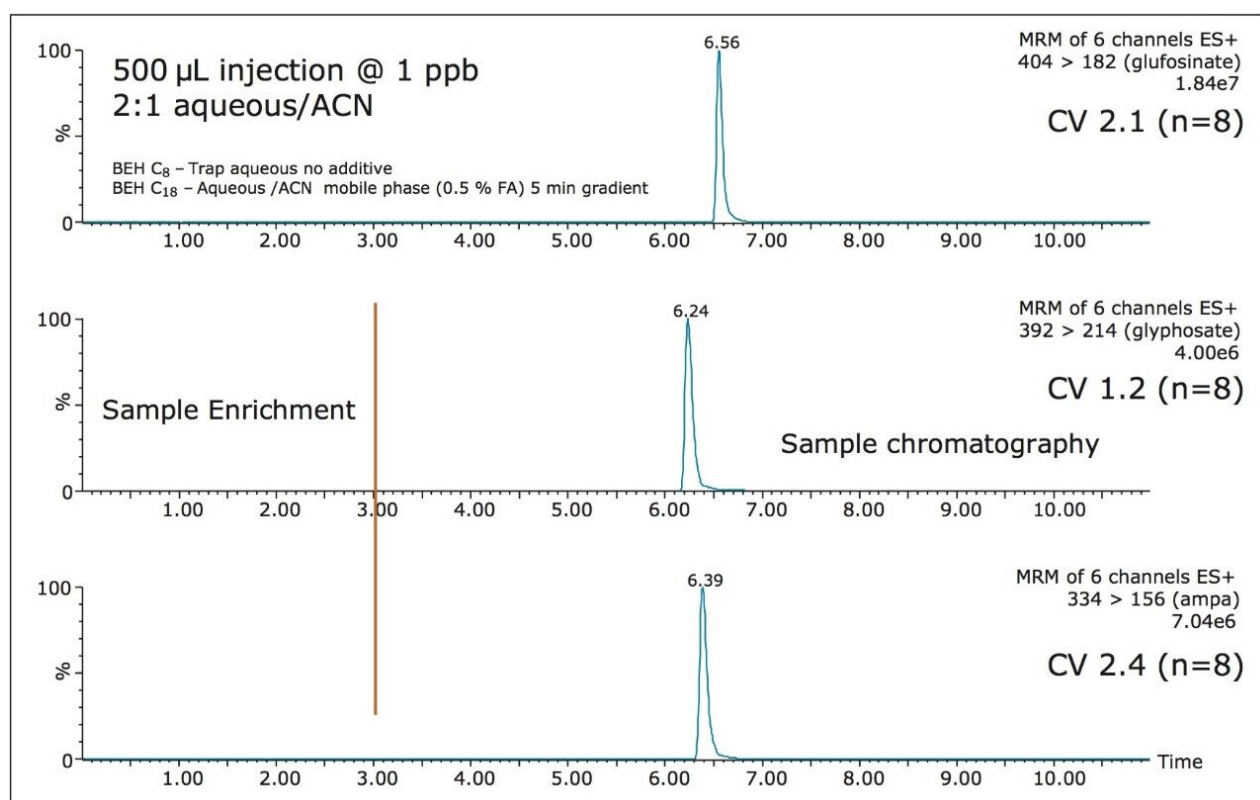


Figure 3. Chromatography profile of FMOc-glyphosate, FMOc-glufosinate, and FMOc-AMPA with at-column dilution active.

### Linearity and quantification

The linearity of the FMOc derivative for glyphosate, glufosinate, and AMPA was measured between 1 ppb and 200 ppb with a 1/x weight and showed an  $r^2$  value of 0.996, 0.993, and 0.991, respectively. The quantification of tap and surface water sample were measured against a MilliQ filtered water calibration curve. The tap and surface water samples were pre-filtered with a 0.45  $\mu\text{m}$  nylon filter with no further treatment. The tap water samples gave a positive signal below 1 ppb (LLOQ), thus giving indication of sub-

ppb detection limit capability with this protocol. The surface water samples gave quantified values of 21.8 ppb, 12.8 ppb, and 18.4 ppb for glufosinate, glyphosate, and AMPA, respectively.

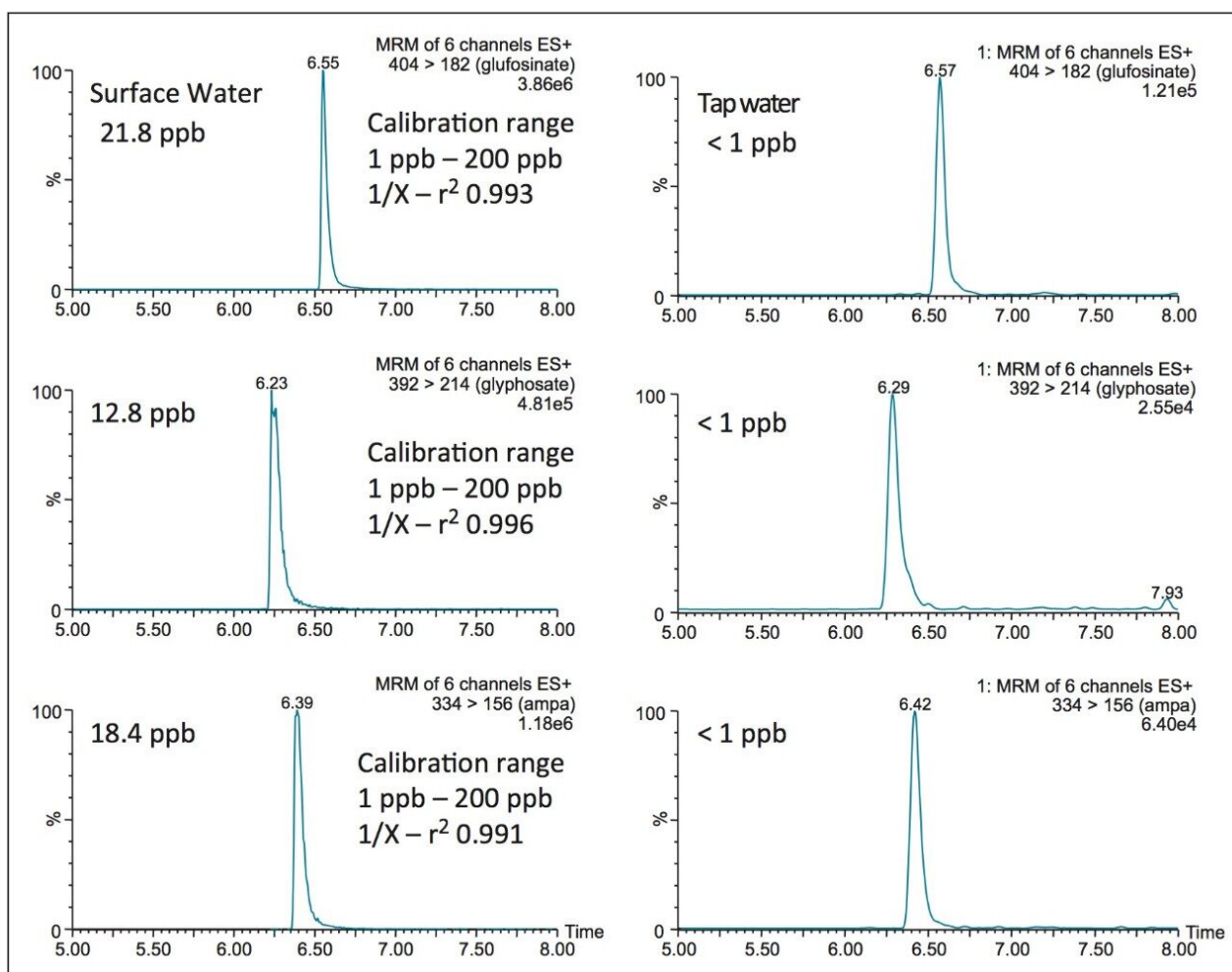


Figure 4. Chromatograms for surface and tap water sample after FMOc derivatization.

## Conclusion

The application targeted the analysis glyphosate, glufosinate, and AMPA in tap and surface water. The limit of quantification in this study was measured at 1.0 ppb. EPA regulations have an MRL in water set at 700 ppb; the 1.0 ppb quantification limit clearly meets these regulations. Since FMOc has UV absorbance properties, similar detection limits could be reached with a photodiode array detector (PDA), thus offering a

cost effective solution.

---

## References

1. Ibanez M, Pozo OJ, Sancho JV, Lopez FJ, Hernandez F, *J. of Chromatogr. A*, 1134, 51, 2006.
2. Martins-Junior HA, Lebre DT, Wang AY, Pires MAF, Bustillos OV, *Rapid Commun. Mass Spectrom.*, 23, 1029, 2009.
3. Hidalgo C, Rios C, Hidalgo M, Salvado V, Sancho J, Hernandez F, *J. of Chromatogr. A*, 1035, 153, 200.
4. Nedelkoska TV, Low GKC, *Analytica Chimica Acta*, 511, 145, 2004
5. Patsias J, Papadopoulou A, Papadopoulou-Mourkidou E, *J. of Chromatogr. A.*, 932, 83, 2001.
6. Mallet, C, Time De-Coupled Chromatography, A Novel Technique, Waters Corporation, 720005125 EN 2014.

---

## Featured Products

ACQUITY UPLC Systems with 2D LC Technology <<https://www.waters.com/10203030>>

720005169, October 2014